

H₂S purification from biogas by direct selective oxidation to sulphur on V₂O₅-CeO₂ structured catalysts

Our research project concerns the H₂S purification from a biogas stream by direct and selective oxidation to sulphur and water at low temperature.

Biogas is a renewable energy source that can be produced from biomass, agricultural and industrial waste and sewage sludge. Its main components are CH₄, CO₂ but unfortunately sulphur based compounds such as H₂S, COS are also present.

The main limitation of the biogas utilization as fuel in Molten Carbonate Fuel Cells is due to the presence of sulphur compounds such as H₂S that can poison the main components of the fuel cell.

The partial H₂S oxidation reaction is carried out in presence of vanadium-based catalysts that are active and very selective toward Sulphur formation.

The aim of the work is to study the preparation procedure of V₂O₅/CeO₂ structured catalysts from the washcoating with CeO₂-ZrO₂ to the deposition of the active phase (V₂O₅) for the abatement of H₂S at 150–200°C.

The catalytic tests were carried out in a fixed bed flow reactor, made of a steel tube 21 cm long and a 14 mm of internal diameter. The reactor is inserted in an electrical furnace equipped with a PID electronic temperature controller. A thermocouple is inserted in a steel sheath concentric to the reactor. The catalytic tests were carried out in the temperature range 150– 250 °C, with a GHSV of 180,000 h⁻¹, by feeding 500 ppm of H₂S, 250 ppm of O₂ and N₂ to balance.

In Figure 1, the scheme of the laboratory plant is shown.

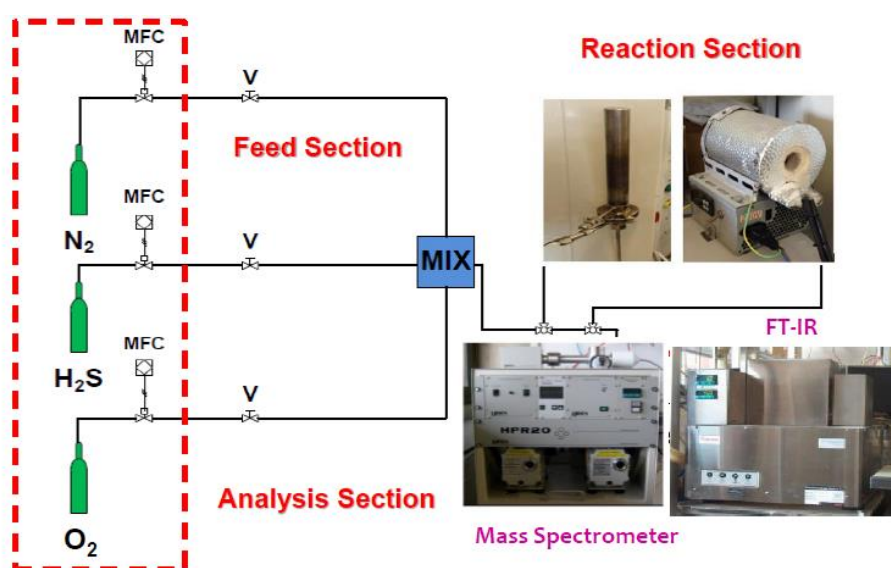


Fig. 1. Laboratory plant for the catalytic tests

The exhaust stream (H₂S, O₂, H₂O) was analyzed by a quadruple filter mass spectrometer (**Hidden HPR-20 QIC R&D**). It was equipped with a sulphur trap to prevent that the sulphur causes the occlusion of the capillary and damage to the fundamental parts of the analyzer.

The concentration of SO₂, which may be present at a very low concentration in the stream leaving the reactor is also monitored by an analyzer FT-IR Multi-gas continually.

In Figure 2 we have shown the typical behaviour of a catalytic activity test carried out by using a catalyst with the highest vanadium loading (19 wt%V₂O₅).

It is worth noting that when the feed stream is sent to the reactor, an immediate and strong decrease of the H₂S concentration and O₂ signal is verified, while a very low SO₂ concentration (10 ppm) is produced. The formation of the water, that remains stable and constant during the test, is indicated by the signal m/z = 18. During this test (2h), no deactivation phenomena were observed. The final H₂S conversion value was higher than 90%.

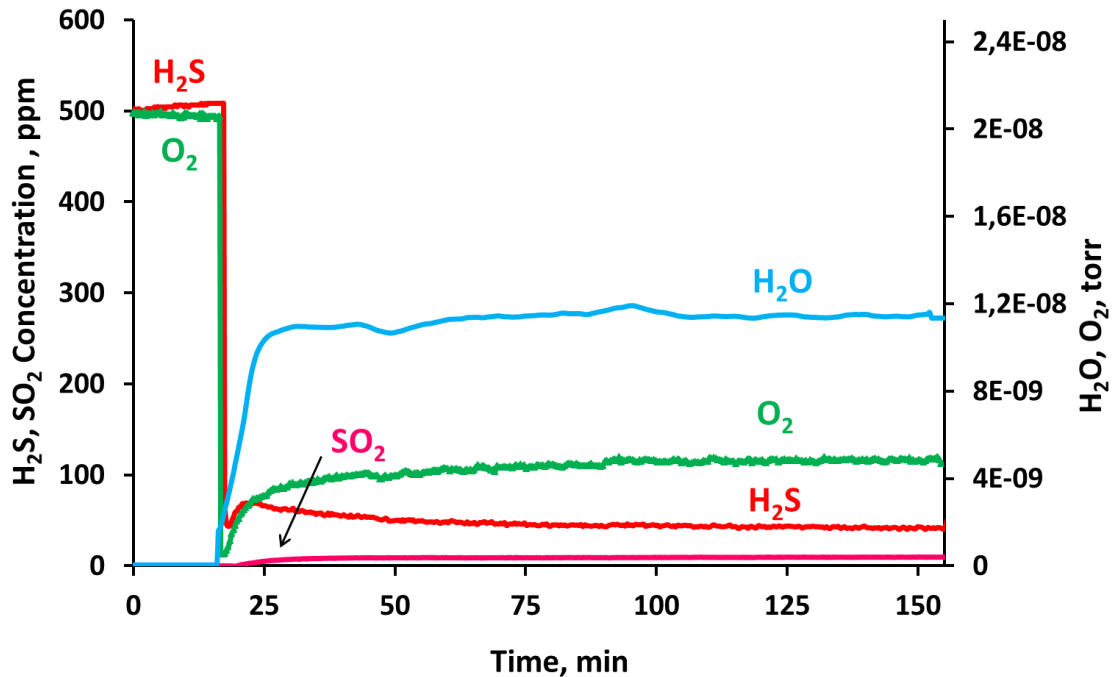


Fig.2. Catalytic activity test of structured catalyst at $T = 200^{\circ}\text{C}$

Finally, the structured catalysts can be considered, based on these results, promising and innovative catalytic systems for the H₂S partial oxidation at low temperatures, having shown good catalytic performances with results very similar to those showed by powder catalysts.

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